

Toxicologie, impact sociétal et transdisciplinarité

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Louvain centre for Toxicology and Applied Pharmacology

SFT, Lille, 27 octobre 2018

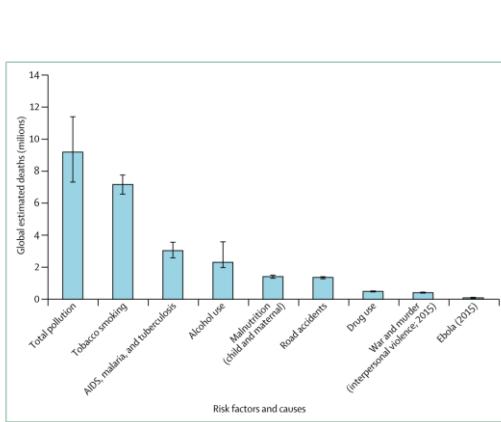


Figure 5: Global estimated deaths by major risk factor and cause, 2015
Using data from the GBD Study, 2016.⁴¹

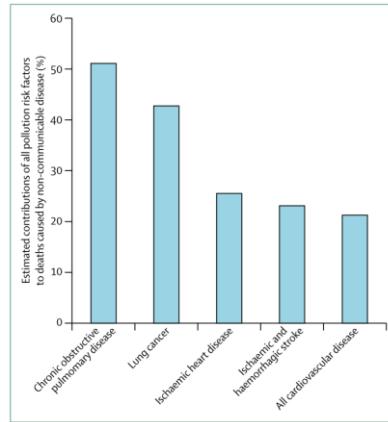
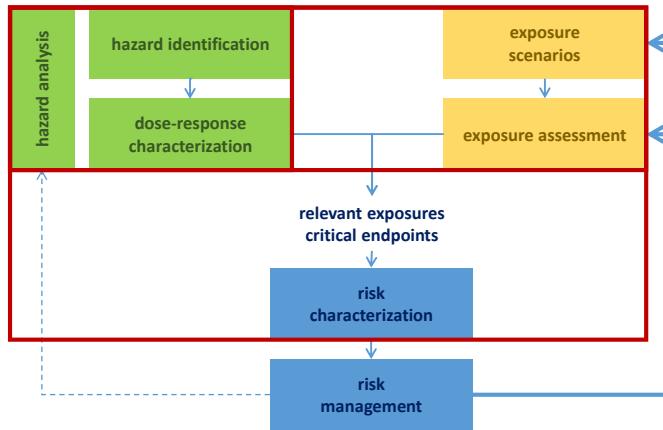
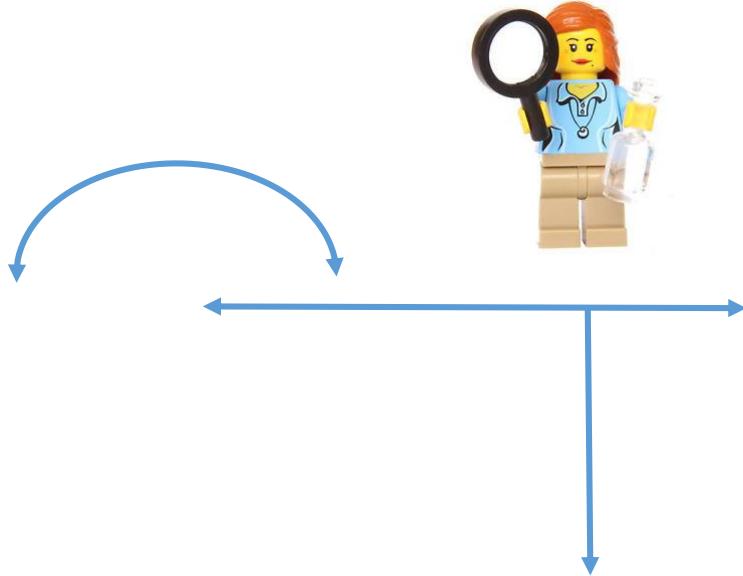


Figure 6: Estimated contributions of all pollution risk factors to deaths caused by non-communicable diseases, 2015
GBD Study, 2016.⁴²

Landrigan et al, 2017

hazard, risk, safe use of chemicals



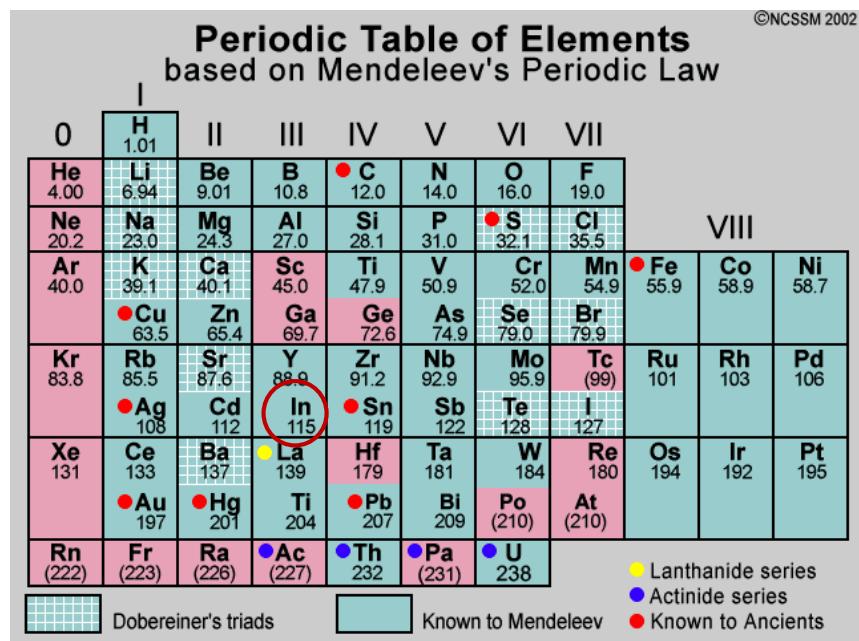


Le toxicologue, Mendeleev et le téléphone

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ITO a new pneumotoxic entity

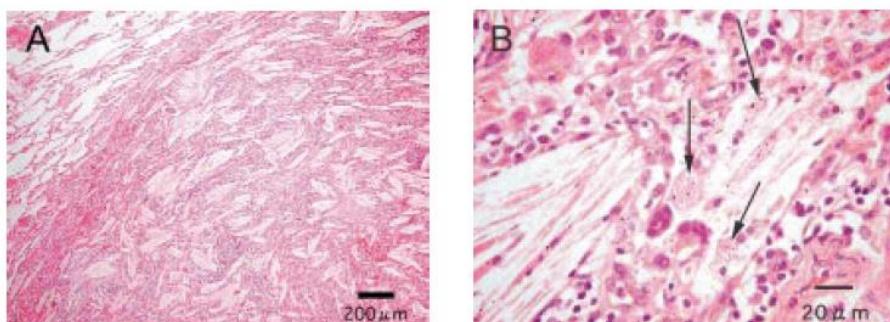


Fig.3. Histopathological sections of lung biopsy, showing interstitial pneumonia. Alveolar spaces are filled with red blood cells, fibrin, cholesterol clefts and alveolar macrophages, but few neutrophils. In the interstitial spaces, perivascular lymphocyte and plasma cell infiltration was seen. (A). Numerous fine particles were observed within the alveolar macrophages and the alveolar spaces (B). Stained with hematoxylin-eosin.

T. Homma et al., 2003

In_2O_3 SnO

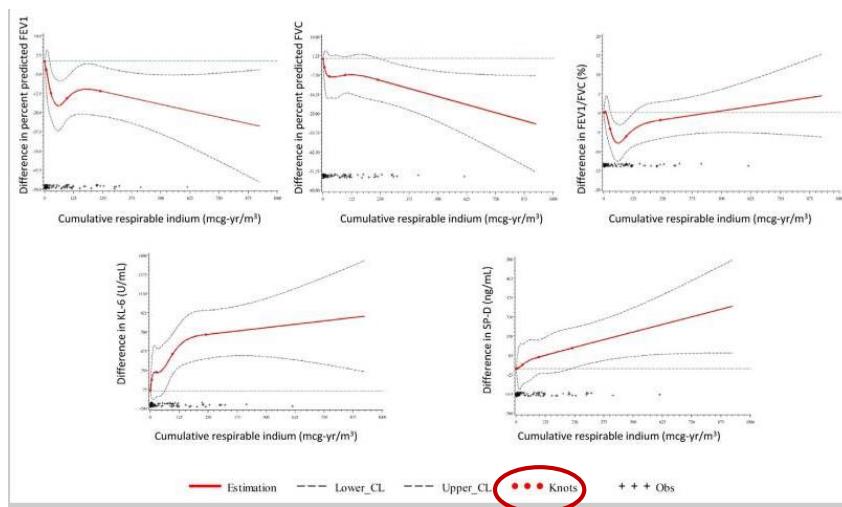
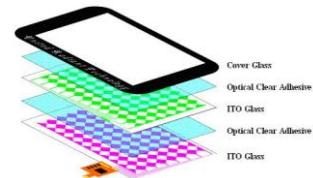
90% 10%

sintering



ITO

sputtering



Cummings et al., 2016



Difference From Reference Values for Pulmonary Function Parameters and Serum Biomarkers by Cumulative Indium Exposure for Workers at an Indium-Tin Oxide (ITO) Facility^a

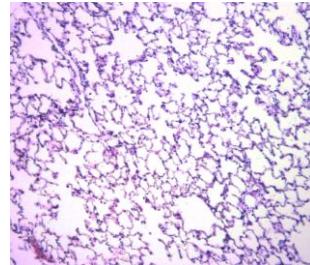
Cumulative indium exposure ($\mu\text{g}\cdot\text{yr}/\text{m}^3$) ^b	Difference (95%CI) ^c				
	FEV1%	FVC%	FEV1/FVC (%)	KL-6 (U/ml)	SP-D (ng/ml)
5.0	-2.4 (-7.7; 2.9)	-3 (-8.3; 2.3)	0.2 (-2.4; 2.8)	98.8 (-61; 259)	1.5 (-50; 53)
12.0	-5.9 (-15.8; 4.1)	-6.2 (-16.1; 3.7)	-0.4 (-5.3; 4.5)	195 (-103.7; 493)	5.3 (-91; 102)
22.0	-10.4 (-20.7; -0.1)	-8 (-18.3; 2.3)	-2.7 (-7.8; 2.3)	222 (-81.8; 530)	13.6 (-85.1; 11)
27.5	-12.5 (-22.2; -2.8)	-8.2 (-17.9; 1.5)	-4.2 (-9; 0.6)	219 (-67.3; 505)	18.5 (-73.7; 111)
30.0	-13.3 (-22.9; -3.8)	-8.3 (-17.8; 1.3)	-4.8 (-9.5; -0.1)	217 (-64.4; 498)	20.7 (-70; 111)
63.0	-17.4 (-27; -7.7)	-8.3 (-17.9; 1.4)	-7.9 (-12.6; -3.1)	285 (2.1; 567)	41.2 (-49.9; 132)
232.0	-11.6 (-21.4; -1.8)	-9.5 (-19.3; 0.3)	-2 (-6.8; 2.8)	662 (376; 948)	92.3 (0.02; 185)
240.0	-11.7 (-21.5; -2)	-9.8 (-19.5; -0.02)	-2 (-6.7; 2.8)	665 (380; 950)	94.6 (2.7; 187)
300.0	-12.9 (-22.5; -3.4)	-11.5 (-21.1; -2)	-1.4 (-6.1; 3.3)	684 (407; 965)	111 (20.5; 202)

Cummings et al., 2016

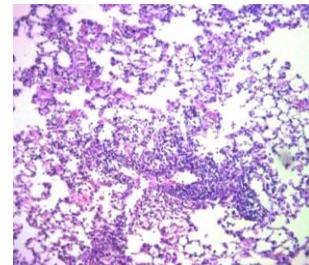
Histopathology (day 60) : alveolar level



NaCl 0.9%
10x

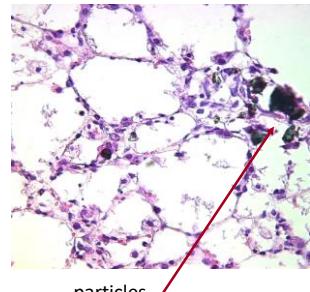


ITO – 2 mg
10x



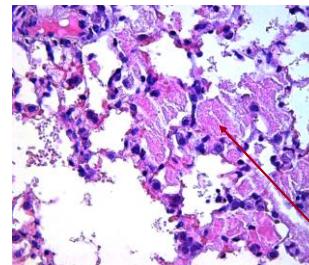
Alveolitis

ITO – 20 mg
40x



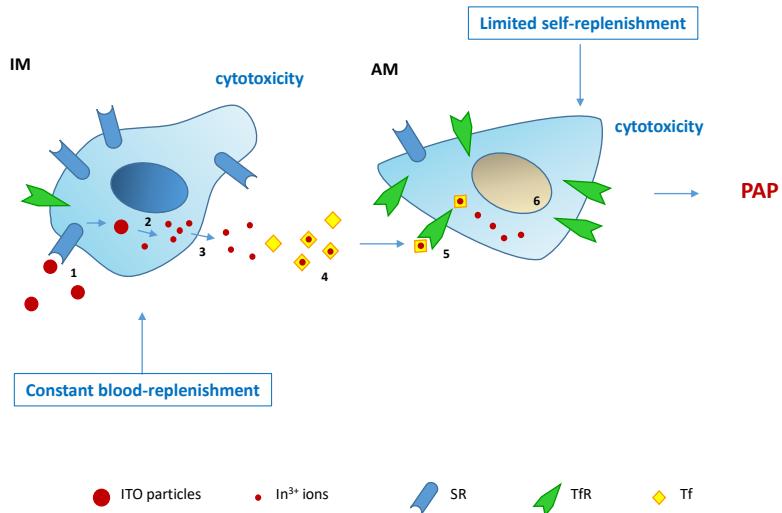
particles

ITO – 2 mg
40x



PAP

Lison et al., 2013



Huaux et al., 2018

Periodic Table of Elements
based on Mendeleev's Periodic Law

©NCSSM 2002

I	II	III	IV	V	VI	VII	VIII
0 He 4.00	Li 6.94	Be 9.01	B 10.8	C 12.0	N 14.0	O 16.0	F 19.0
Ne 20.2	Na 23.0	Mg 24.3	Al 27.0	Si 28.1	P 31.0	S 32.1	Cl 35.5
Ar 40.0	K 39.1	Ca 40.1	Sc 45.0	Ti 47.9	V 50.9	Cr 52.0	Mn 54.9
	● Cu 63.5	Zn 65.4	Ga 69.7	Ge 72.6	As 74.9	Se 79.0	Br 79.9
Kr 83.8	Rb 85.5	Sr 87.6	Y 88.9	Zr 91.2	Nb 92.9	Mo 95.9	● Fe 55.9
	● Ag 108	Cd 112	In 115	● Sn 119	Sb 122	Te 128	Co 58.9
Xe 131	Ce 133	Ba 137	● La 139	Hf 179	Ta 181	W 184	Ru 101
	● Au 197	● Hg 201	Ti 204	● Pb 207	Bi 209	Po (210)	Rh 103
Rn (222)	Fr (223)	Ra (226)	● Ac (227)	● Th 232	● Pa (231)	● U 238	Pd 106
Dobereiner's triads		Known to Mendeleev				Lanthanide series	



Lung Function Changes in Workers Exposed to Cobalt Compounds

A 13-Year Follow-up

TABLE 2. RANDOM COEFFICIENTS MODEL FOR FEV₁* AND FVC†

Effect	FEV ₁			FVC		
	Estimate	SE	p Value	Estimate	SE	p Value
Smoking						
Intercept	0.1887	0.0083	< 0.001	0.2086	0.0106	< 0.001
Age at baseline	-0.0017	0.0003	< 0.001	-0.0015	0.0003	< 0.001
Log Co-U	-0.0024	0.0011	0.026	-0.0024	0.0014	NS
Time	-0.0016	0.0002	< 0.001	-0.0019	0.0004	< 0.001
Nonsmoking						
Intercept	0.1688	0.0091	< 0.001	0.1921	0.0092	< 0.001
Age at baseline	-0.0010	0.0003	< 0.001	-0.0010	0.0003	< 0.001
Log Co-U	-0.0013	0.0011	NS	-0.0003	0.0014	NS
Time	-0.0017	0.0002	< 0.001	-0.0017	0.0002	< 0.001

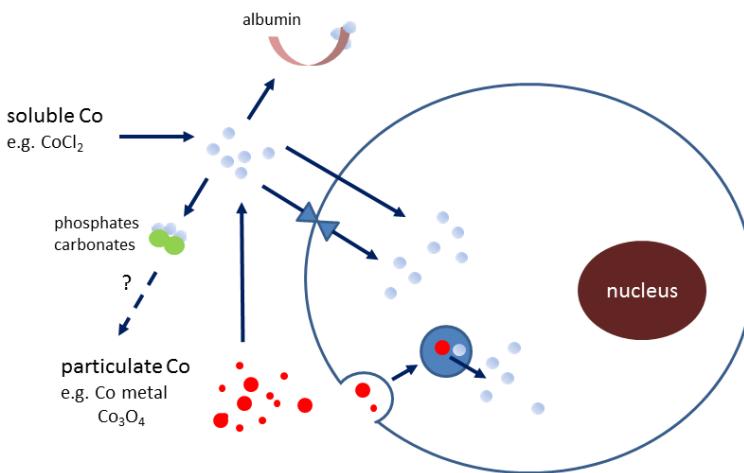


Verougstraete et al., 2004

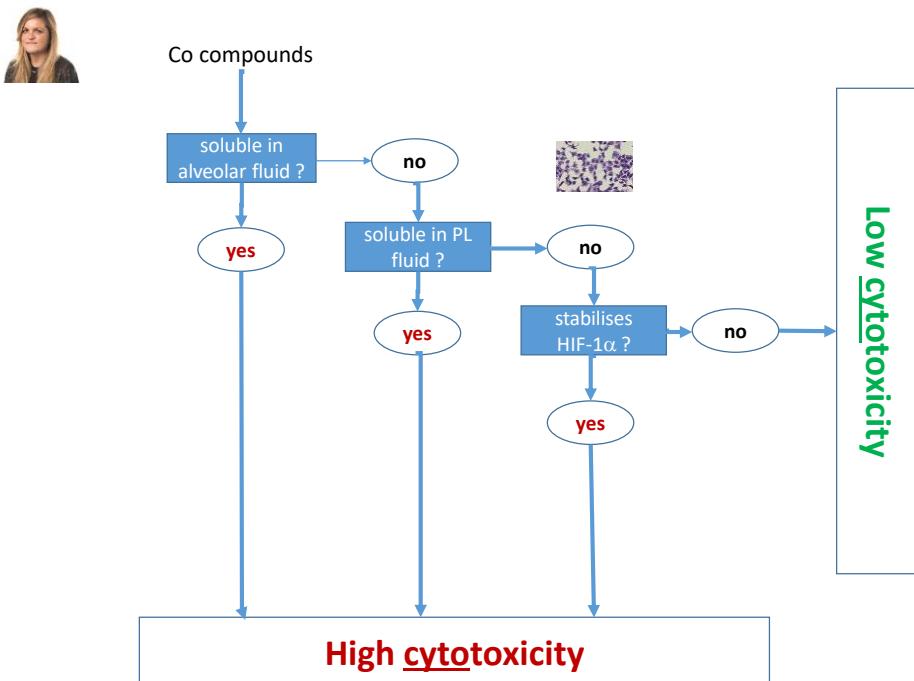
Cobalt compounds

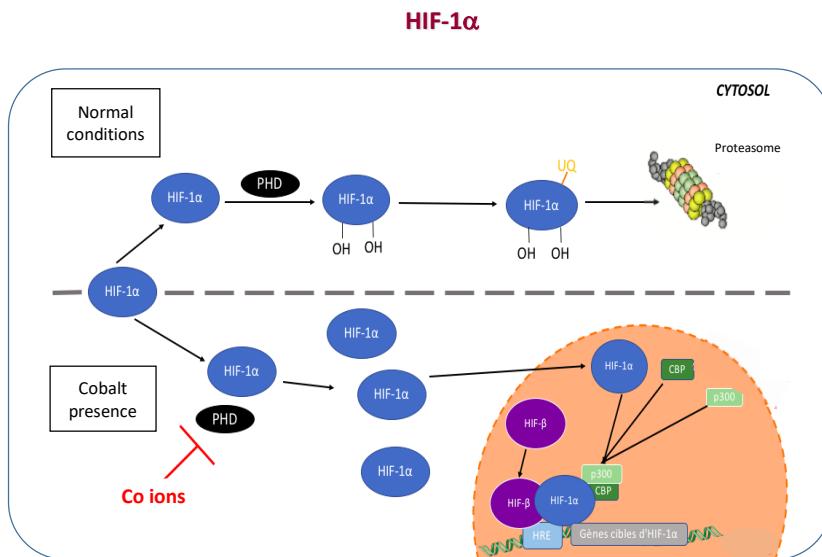
- Cobalt metal
- Cobalt chloride
- Cobalt sulfate
- Cobalt (II) oxide
- Cobalt (II,III) oxide
- Cobalt hydroxide
- Cobalt sulfide
- Cobalt octoate
- Cobalt naphtenate
- ...

*Grouping ?
Read across ?*

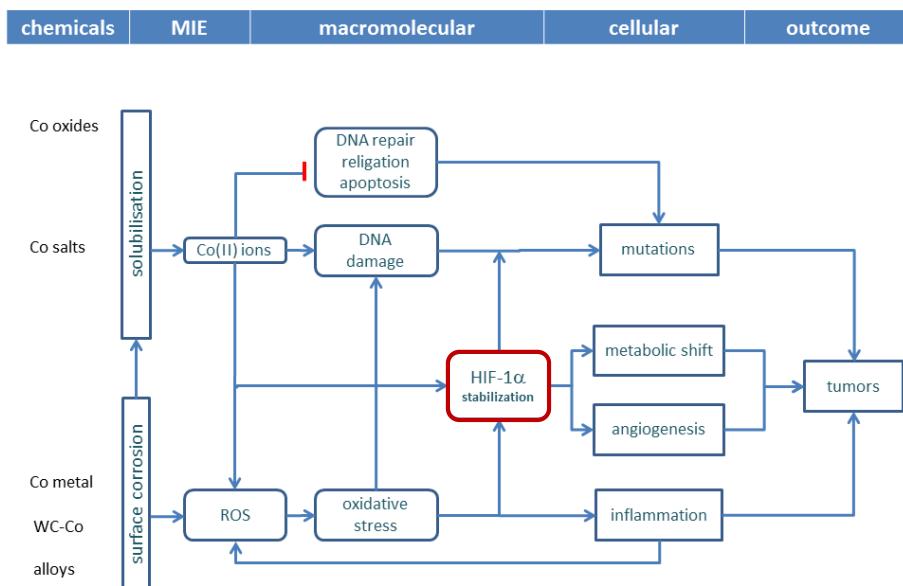


Lison et al., 2018





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Lison et al., 2018

©NCSSM 2002

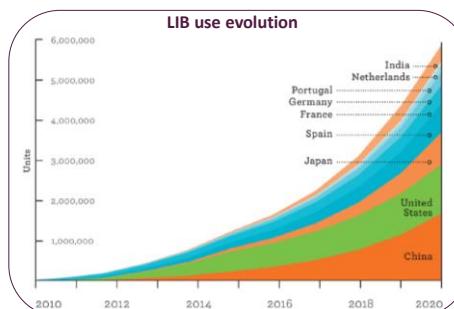
Periodic Table of Elements based on Mendeleev's Periodic Law

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Xe 131	Ce 133	Ba 137	● La 139	Hf 179	Ta 181	W 184	Re 180	Pt 195
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Rn (222)	Fr (223)	Ra (226)	● Ac (227)	● Th 232	● Pa (231)	● U 238		

Dobereiner's triads
 Known to Mendeleev
 ● Known to Ancients



Li-ion battery (LIB)



LIB particles
 LiFePO₄ (**LFP**), LiCoO₂ (**LCO**), LiMn₂O₄ or Li₂MnO₃ (**LMO**),
 Li₄Ti₅O₁₂ (**LTO**), LiNiMnCoO₂ (**NMC**), LiNiCoAlO₂ (**NCA**)



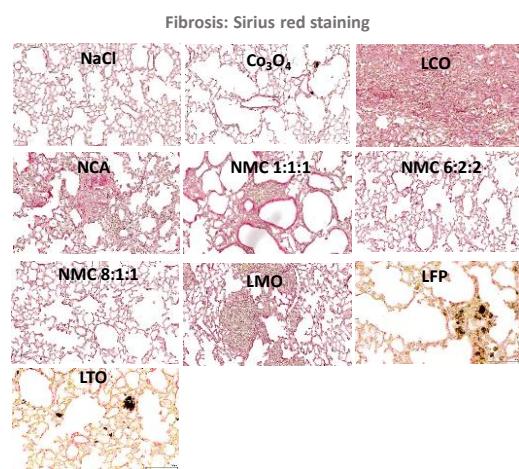
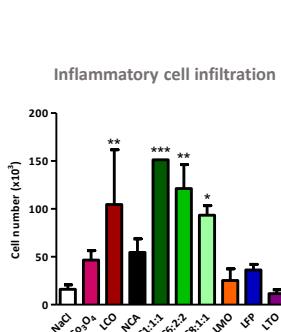
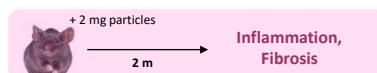
Is lung toxicity due to Co? Comparison of the lung toxicity of a large range of LIB particles

Co_3O_4	73 % Co
LiCoO_2 (LCO)	60 % Co
LiNiCoAlO_2 (NCA)	27 % Co
CoLiNiMnCoO_2 (NMC)	
1:1:1	20 % Co
6:2:2	12 % Co
8:1:1	6 % Co
LiFePO_4 (LFP)	
$\text{Li}_4\text{Ti}_5\text{O}_{12}$ (LTO)	
LiMn_2O_4 or Li_2MnO_3 (LMO)	

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Is lung toxicity due to Co? Comparison of the lung toxicity of a large range of LIB particles



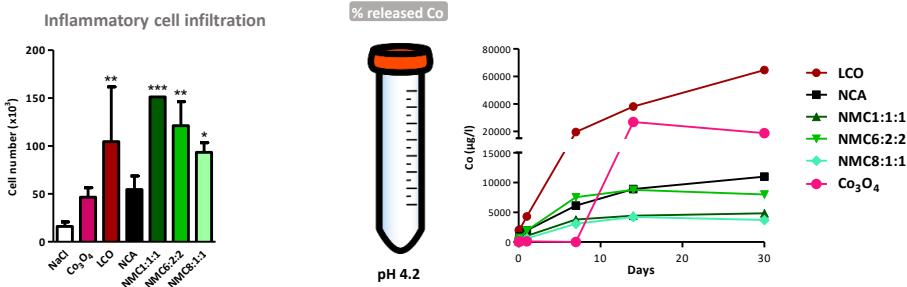
→ No association with Co content

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Is lung toxicity due to Co?

Comparison of the lung toxicity of a large range of LIB particles



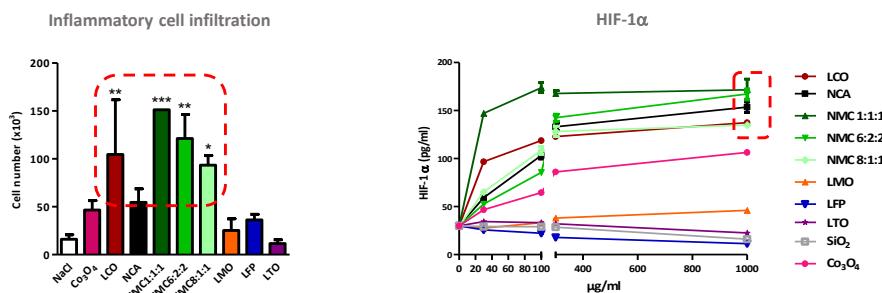
→ No association between Co %, Co bioaccessibility and LIB lung toxicity

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Is lung toxicity due to Co?

Comparison of the lung toxicity of a large range of LIB particles

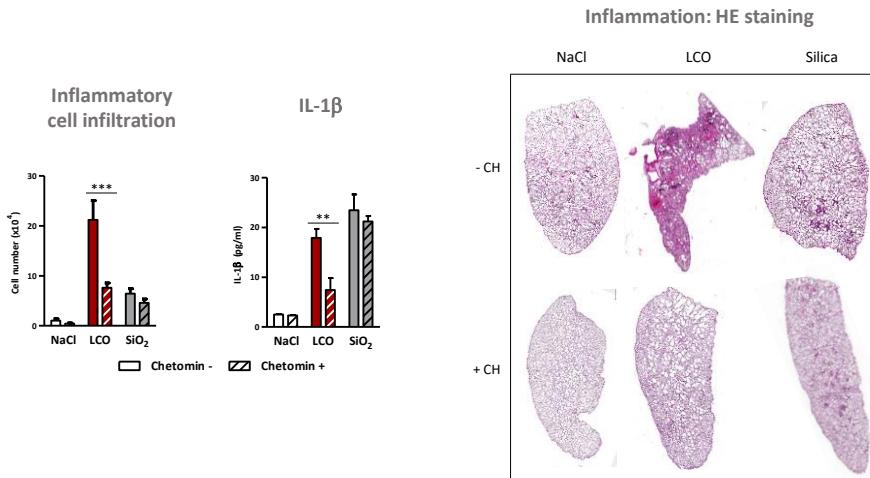


→ Association with biologically active Co (and other transition metals)

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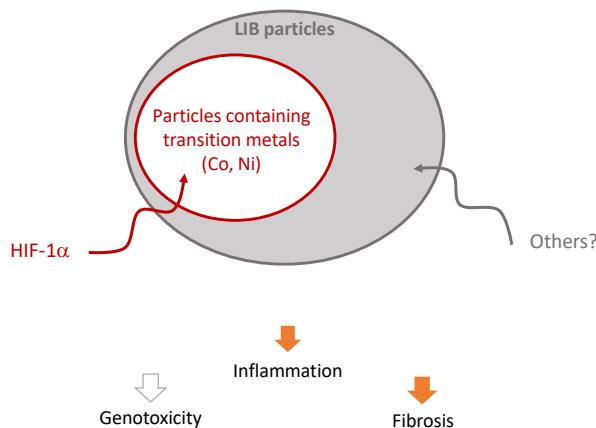
Implication of HIF-1 α in LCO toxicity:



→ HIF-1 α is implicated in LCO lung toxicity
 → HIF-1 α could represent a key event of LIB particle containing Co and/or transition metals



Societal usefulness



Periodic Table of Elements
based on Mendeleev's Periodic Law

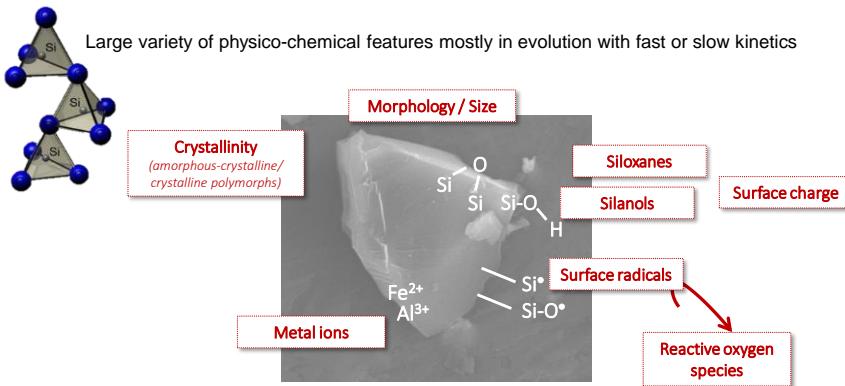
©NCSSM 2002

0	H 1.01	II	III	IV	V	VI	VII	
He 4.00	Li 6.94	Be 9.01	B 10.8	C 12.0	N 14.0	O 16.0	F 19.0	
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Dobereiner's triads
 Known to Mendeleev
 ● Known to Ancients

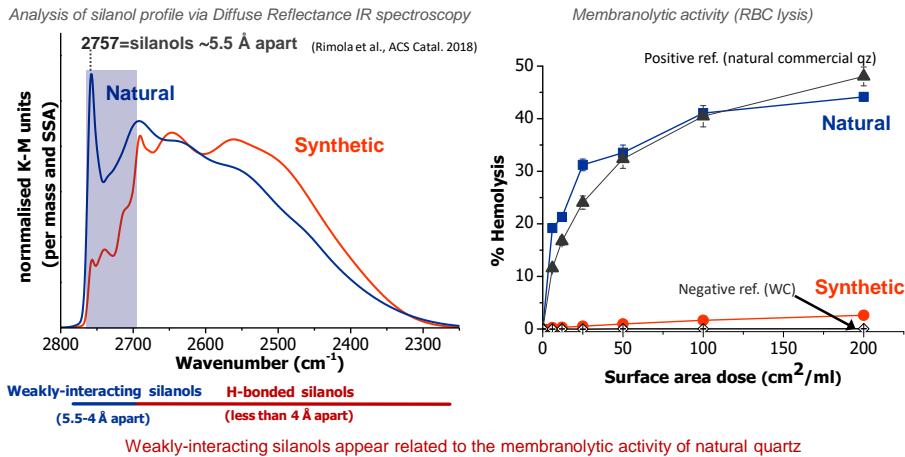


Many physico-chemical properties of silica may impart toxicity





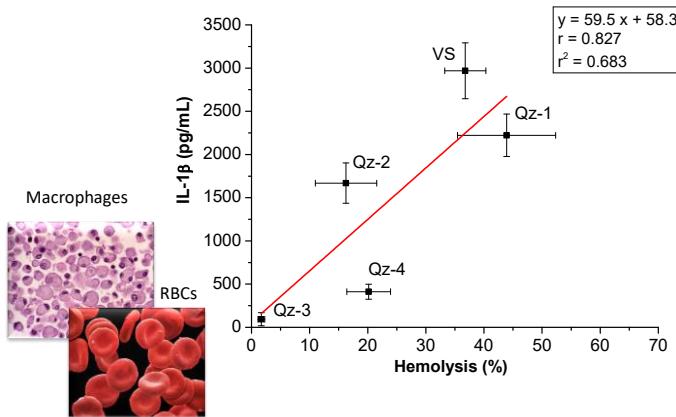
Synthetic vs natural quartz: silanol arrangements & membranolysis



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IL-1 β release correlates with silica hemolytic activity

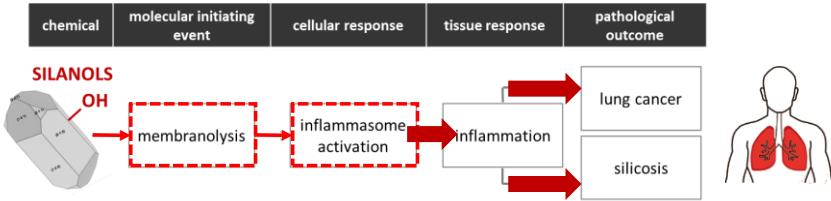


Pavan et al., Part & Fibre Tax, 2014

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Proposed adverse outcome pathway (AOP) for silicosis and lung cancer



Pavan & Fubini, Chem. Res. Toxicol. 2017

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Messages

- → Impact sociétal
 - descriptive
 - dose-réponse
 - mécanisme ... KE ... PoD ... acceptable exposure
- Trans-disciplinarité
 - in vitro alone ?
 - Tox et Ecotox (microbiote)
 - Hors toxicologie
- Un rôle pour
 - académiques
 - sociétés scientifiques

Formation continue #

Toxicology

IUFC

[Retour à la page supérieure](#)

Registration Form

Target Audience and Conditions

Objectives

Programme

Teaching staff

Teaching and Certification



Toxicology

Registrations are open!

The Faculty of Public Health at the Catholic University of Louvain (UCL) and the Belgian Society of Toxicology and Ecotoxicology (BelTox) jointly organise a new university certificate in "Toxicology".

This continuing education programme will consist of three modules in 2019. Additional modules will be offered in the years to come. Courses are organised into one-week modules scheduled between January and December 2019 in Louvain-la-Neuve.